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Kenneth S. Krane is Professor of
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where he has served on the faculty since 1974, including 14 years as Department Chair. He received the Ph.D. in nuclear physics from Purdue University in 1970 and held postdoctoral research positions at the Los Alamos National Laboratory and the Lawrence Berkeley National ...

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The total m_s is then $5 \times (+1/2) +$
 $(-1/2) = 2$. (c) The five electrons with
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 m_l values $(+2, +1, 0, -1, -2)$, which

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sum to zero. The sixth electron has a maximum m_l of +2, so the maximum total m_l is +2. (d) The next available level for one of the 3d electrons is 4p. The remaining five 3d.

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